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## INFORMATION RETRIEVAL\*

346, 389, 646

Key Words: A. Kinetics-8, Correlation-2, Computers-10, Differential Equations-1, Concentrations-2, Integration-10. B. Kinetics-8, Ethylene Oxide-1, Ammonia-1, Ethanolamines-2, Benzene-1, Chlorine-1, Chlorobenzene-2, Methanol-1, Propylene-1, Proplyene Oxide-1, Correlation-1, Computers-10, Rate Constants-2, Concentrations-1, Least Squares-4.

Abstract: A new method of obtaining an approximate integral solution of a set of differential rate equations is described. The method is used to correlate experimental data on systems whose stoichiometry would indicate a consecutive competitive mechanism. The estimates of the rate constants, found by fitting the approximate solution to the data, are within experimental error of the values obtained by other methods.

Reference: Friedman, M. H., and R. R. White, A.I.Ch.E. Journal, 8, No. 5, p. 581 (November, 1962).

\* For details on the use of these key words and the A.I.Ch.E. Information Retrieval Program, see Chem. Eng. Progr., 57, No. 5, p. 55 (May, 1961), No. 6, p. 73 (June, 1961); 58, No. 7, p. 9 (July, 1962).

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#### ERRATUM

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It has been called to the attention of R. E. Emmert and R. L. Pigford by Professor P. V. Danckwerts that some data which they extracted from a paper by Bates and Pinching® were misinterpreted in "Gas Absorption Ac-companied by Chemical Reaction," which appeared in the May, 1962, issue of the A.I.Ch.E. Journal. As a consequence, Emmert and Pigford's analvsis of the relative effects of competing reactions based on previous workers' data must be modified. This change does not in any way alter their own data or their interpretation. This erratum describes the changes needed to account for this misinterpretation. All changes are in the section labeled "Reaction Mechanism."

In order to determine the concentrations of hydroxyl ion and free amine

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<sup>&</sup>lt;sup>o</sup> Bates, R. G., and G. D. Pinching, J. Research Natl. Bur. Standards, 46, 349 (1951).

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